respollution Bulletin, Vol. 11, pp. 18-22 hymon Press Ltd. 1980. Printed in Great Britain.

# Mercury in Water, Organisms and Mediments from a Section of the Turkish Mediterranean Coast

TUNCEL,\* G. RAMELOW\* AND T. I. BALKAS, parlment of Marine Sciences, Middle East Technical University, P. K. 28, Erdemli/Içel, Turkey (hoom address: Department of Chemistry, Middle East Technical University, Ankara, Turkey.

mes of water, marine organisms and sediment collected in the Turkish Mediterranean coast in the vicinity of this were analysed for total mercury. The levels of the true of true of the true of the true of the true of true of the true of true o

hamful effects of toxic chemicals to the marine of the particularly in coastal areas, are well known. I pographical area of particular concern is the Mediterraments. This essentially closed basin is bordered by many parties, some of which are highly developed industrially. The is a large input of polutants into the Mediterranean town land-based sources (Helmer, 1977). Until recently no tempts were made to control the input of toxic chemicals the sea and as a result severe pollution problems, especitly in localized areas, have been observed (Osterberg & Leis 1977).

in the wake of the world-wide concern over the harmful stomental effects of mercury, several investigations of metals of mercury in the Mediterranean marine environmental eben carried out in recent years. (Thibaud, 1971; traciolo et al., 1972; Robertson et al., 1972; Renzoni et al., 1973; Selli et al., 1973; Roth & Hornung, 1975; Fukai & trab Ngoc, 1976; Grimanis et al., 1976). These demonstrated that although in general the read mercury are similar to those found in other regions the world, elevated levels are seen in certain coastal areas.

The study reported here was carried out to make a survey of the levels of mercury in the marine environment in an area of the eastern Mediterranean along the Turkish coast comprising the Mersin harbour area and the region to the west of Mersin.

## Materials and Methods

Water

Water samples were collected at three locations, as shown in Fig. 1. At Ovacik samples were collected at five stations; at Limonlu at three stations; and at Mersin at nine stations, both inside and outside the harbour. The samples were collected over 1 year at approximately 2-monthly intervals at Ovacik; Limonlu and Mersin samples were collected only twice.

The samples were taken with 1.51. Nansen or 2.51. Niskin plastic water samplers. Immediately after collection they were transferred to 11. polyethylene bottles. A preservative solution consisting of 10 ml 2% KMnO<sub>4</sub> and 10 ml concentrated H<sub>2</sub>SO<sub>4</sub> was added to the samples within 3 h of collection.

Due to the extremely low levels of mercury in unpolluted seawater, a concentration step was needed before analysis. The permanganate enrichment technique (Topping & Pirie, 1972; Harsanyi et al., 1973) was used. Samples concentrated 80 times were analysed by atomic absorption spectrometry using a cold-vapour apparatus of our own design.

To study the loss of mercury from seawater samples during storage, 25 1-1. seawater samples were collected from a selected location. The samples were divided into five

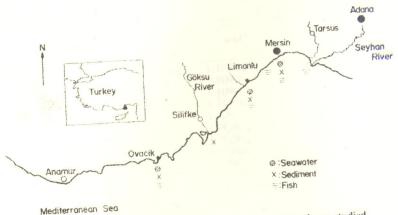


Fig. 1 Sampling locations in eastern Mediterranean coastal area studied.

goups, each group containing five samples. To each sample athe first group were added 50 ml HNO3; to each in the group, 50 ml HNO<sub>3</sub>+10 ml 2% K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>; to each in the third group 10 ml H<sub>2</sub>SO<sub>4</sub>+10 ml 2% KMnO<sub>4</sub>; to the burth group samples no additions were made. The amaining five samples were analysed within 24 h. The four goups of treated samples were stored for 6 months and then analysed.

# **Organisms**

Seven different species of fish and one species of shrimp are analysed. The organisms were collected mainly near the coast between Mersin and Limonlu; a few samples from the Seyhan Delta were also analysed. Specimens were allected both by deep trawl and with gill nets. After allection the samples were taken in an ice box to the aboratory where they were positively identified and the legths and weights recorded. Each specimen was stored in a plastic bag at -40° Cuntil analysis.

For analysis 0.5-1 g of muscle tissue was carefully rmoved from the dorsal area (or abdomen in the case of the drimp) and digested with 3 ml HNO3 at 140°C for 3 h in Tellon-lined high-pressure decomposition vessels. After aution the digested samples were analysed by cold-vapour atomic absorption spectrometry.

# Sediment

Sediment samples were collected at four locations, as shown in Fig. 1, with two stations at each location. All amples were collected with a Van Veen type grab sampler. After collection the samples were placed in plastic bags and stored wet at - 40°C.

For analysis the frozen sediment was thawed and 5-7 portions in weights of 0.05-0.3 g (wet) were selected from each sample. Each portion was digested with 2 ml HNO<sub>3</sub> in high-pressure decomposition vessels at 140°C for 9 h. After dution the digested samples were analysed by cold-vapour AAS. All samples were analysed wet to avoid any mercury lusses during drying. To express analytical results on a drywight basis, the samples were dried at 110°C for 24 h and the percentage of water calculated.

To study the effect of acid mixture on the extraction of necury from marine sediments, a sample was dried and 0.05 g portions digested with different acids and acid mixtures at 150°C for 5 h. The diluted samples were then analysed for their mercury content. To determine the dependence of extraction efficiency on digestion time, 0.05 g

portions of the same sediment sample were digested with 2 ml HNO<sub>3</sub> for periods of time from 1 to 9 h.

# Results and Discussion

The average concentrations of mercury in the water samples analysed are presented in Table 1. The values in this table from Ovacik are the averages of the samples taken during the last three sampling periods because the initial samples collected were used to develop the procedure as a routine method. The mercury concentrations given for each station at Limonlu and Mersin are the averages of two sample collections. The number of replicate analyses at each station does not allow computation of standard errors, but ten replicate analyses of a single water sample showed a relative standard deviation of about 10%, covering the entire procedure from enrichment to analysis steps.

The data in Table 1 indicate that the levels of mercury in water samples from the region of the Mediterranean studied are low. The samples from Mersin harbour show mercury concentrations 2-3 times higher than the samples collected at Ovacik and Limonlu, although still comparable with the levels found in other regions of the Mediterranean (UNEP, 1977). No differences in mercury levels between surface and bottom samples was observed.

Due to the low concentrations of mercury in unpolluted seawater, a problem of storage arises. If the samples must be kept for several days, either on board ship or in the laboratory, before analysis, it is necessary to ensure that no loss of mercury occurs during this period. The mechanism of mercury loss, although not fully understood, is probably due to both adsorption on the walls of the container and volatilization (Rosain & Wai, 1973). Many different preservatives have been recommended in the literature among them HNO, (Rosain & Wai, 1973), H<sub>2</sub>SO<sub>4</sub>-KMnO<sub>4</sub> mixture (Harsanyi et al., 1973) and HNO<sub>3</sub>-K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> mixture (Christman & Ingle, 1976).

The ability of various preservatives to prevent loss of mercury from seawater samples was tested. The results given in Table 2 indicate that no solution can preserve samples with 100% efficiency, but the HNO<sub>3</sub>-K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> appears to be the best, resulting in only a 20% loss in 6 months. This is in good agreement with the work of Christman & Ingle (1976) who found a 100% retention with the same preservative over a period of 6 days.

TABLE 1 Mercury in seawater at three locations along the Turkish Mediterranean coast.

|  |         | pun berust.  |                                     |  |
|--|---------|--------------|-------------------------------------|--|
| Location   | Station | Depth<br>(m) | Mercury<br>concentration<br>(ng l1) |  |
| Ovacsk   | 1       | 0            | 9                                   |  |
| The second second  | 1       | 3            | 8                                   |  |
|  | 2       | 0            | 8                                   |  |
| The second   | 2       | 5            | 8                                   |  |
|  | 3       | 0 .          | 7                                   |  |
|  | 3       | 50           | 8                                   |  |
| Contract Con | 4       | 0            | 8                                   |  |
|  | 4       | 5            | 10                                  |  |
|  | 5       | 0            | 7                                   |  |
| The second   | 5       | 20           | 8                                   |  |
| Limonlu  | 6       | 0            | 10                                  |  |
|  | 6       | 20           | 12                                  |  |
|  | 7       | 0            | 10                                  |  |
|  | 7       | 20           | 11                                  |  |
| Mersin harbour   | 8       | 0            | 11                                  |  |
| MOSIII HALDOUL   |         | 20           | 24                                  |  |
|  | 10      | 10           | 27                                  |  |
|  | 11      | 20           | 23                                  |  |
|  | 12      | 10           | 23                                  |  |
|  | 13      | 20           | 23                                  |  |
|  | 14      | 10           | 40                                  |  |
|  | 15      | 5            | 21                                  |  |
|  | 16      | 5            | 24                                  |  |
|  | 17      | 10           | 19                                  |  |

TABLE 2 Loss of mercury from seawater samples during storage.\*

| Preservative     | Mercury con | % mercury lost |   |
|------------------|-------------|----------------|---|
|                  | Initial     | Final          | , |
| None             | 10          | 1              | 90                                      |
| HNO <sub>3</sub> | 10          | 3              | 70                                      |
| HSO4:KMnO4       | 10          | 5              | 50                                      |
| HNO3:K2Cr2O7     | 10          | 8              | 20                                      |

'All samples were stored for 6 months.

TABLE 3 Total mercury (µg kg<sup>-1</sup> wet wt) in some marine organisms.

| Species             | Length (mm) | No. of specimens | Range   | Mean |
|---------------------|-------------|------------------|---------|------|
| Mullus barbatus     | 98-192      | 9                | 20-80   | 40   |
| Mullus surmuletus   | 150-175     | 5                | 40-80   | 70   |
| Boops salpa         | 187-204     | 4                | 2-4     | 3    |
| Mugil auratus       | 241-295     | 5                | 10-30   | 20   |
| Upeneus moluccensis | 135-148     | 5                | 110-130 | 120  |
| Sardina pilchardus  | 157-165     | 5                | 40-60   | 50   |
| Seriola dumeritii   | 181         | 1                | raper   | 100  |
| Peneus kerathurus   | 168-170     | 6                | 10-40   | 20   |

Total mercury was determined in several different marine organisms, some of economic importance in the weastudied. As seen from the data in Table 3, the levels of mercury in the organisms studied were low in all cases. The highest concentrations of mercury were found in a species of fish, Upeneus moluccensis. The mercury levels in marine organisms in the area studied are considerably lower than those reported along the Israel coast (Roth & Hornung, 1975).

# Sediment

To ensure complete extraction of mercury from the ediment samples, several different acids and acid mixtures

TABLE 4 Extraction of mercury from a marine sediment.\*

| ml: acid                         | No. of   | Merci<br>(µg g | 0711          |                      |  |
|----------------------------------|----------|----------------|---------------|----------------------|--|
| go.                              | analyses | Mean           | Std deviation | % relative std. dev. |  |
| 1:HNO3                           | 11       | 0.65           | 0.05          | 7.7                  |  |
| 2:HNO <sub>3</sub>               | 5        | 0.60           | 0.04          | 6.7                  |  |
| 3:HNO,                           | 5        | 0.56           | 0.02          | 3.6                  |  |
| 4:HNO,                           | 5        | 0.54           | 0.02          | 3.7                  |  |
| 5:HNO,                           | 5        | 0.56           | 0.02          | 3.6                  |  |
| 1:H <sub>2</sub> SO <sub>4</sub> | 4        | 0.16           | 0.02<br>t     | 3.0                  |  |
| 3:H2SO4                          | 4        | 0.38           | _ +           | -                    |  |
| 5:H <sub>2</sub> SO <sub>4</sub> | 5        | 0.45           | 0.03          | . 7                  |  |
| 1:HNO3:0.5 HCIO4                 | 6        | 0.50           | 0.03          | 6.7                  |  |
| 1:HNO1:1 HClO4                   | 6        | 0.59           | 0.04          | 6.8                  |  |
| 1:HNO3:1 HF                      | 7        | 0.49           | 0.06          | 12.2                 |  |
| 1:HNO <sub>3</sub> :3 HF         | 7        | 0.54           | 0.12          | 22.2                 |  |
| 2:HNO3:H2SO4                     | 16       | n.d.           | 0.12          | Andre da             |  |

\*0.05 g dry sediment was used in all cases.

The number of analyses was not sufficient to calculate the standard deviation.

n.d. - none detected.

TARIF 5 Effect of digestion period on extraction of mercury from sediments.\*

|      | Mercury found |         |            |          |  |
|------|---------------|---------|------------|----------|--|
| Time | No. of        | (µg g - | % relative |          |  |
| (h)  | analyses      | Mean    | Std dev.   | std dev. |  |
| 1    | 8             | 0.62    | 0.07       | 11.3     |  |
| 3    | 6             | 0.60    | 0.04       | 6.7      |  |
| 5    | 7             | 0.54    | 0.05       | 9.3      |  |
| 7    | 7             | 0.58    | 0.03       | 5.2      |  |
| 9    | 7             | 0.59    | 0.02       | 3.4      |  |

\*0.05 g dry sediment was digested with 2 ml HNO1.

TABLE 6 Mercury in sediments taken from four locations along the Turkish Mediterranean coast

| C1                |    | Mercury concentration (µg g -1) |               |  |
|-------------------|----|---------------------------------|---------------|--|
| Sampling location | N* | Mean                            | Std deviation |  |
| Ovacik            | 5  | 0.019                           | 0.006         |  |
|                   | 5  | 0.036                           | 0.005         |  |
| Goksu delta       | 6  | 0.037                           | 0.004         |  |
|                   | 5  | 0.058                           | 0.008         |  |
| Limonlu           | 6  | 0.022                           | 0.002         |  |
|                   | 6  | 0.034                           | 0.004         |  |
| Mersin harbour    | 6  | 0.44                            | 0.03          |  |
|                   | 7  | 0.48                            | 0.04          |  |

\*N refers to the number of subsamples analysed from each grab.

were evaluated for their efficiency in extracting mercury from a selected sediment sample. The results summarized in Table 4 indicate that HNO, alone is quite sufficient for extracting mercury from marine sediments of this type and leads to the best precision in replicate analysis of a single sample. A surprising result was the discovery of the absence of any mercury absorption for the sample digested with HNO<sub>3</sub>-H<sub>2</sub>SO<sub>4</sub> mixture. In 16 replicate analyses with this commonly used mixture no mercury was found. A large non-specific background absorption was observed, however, during the analysis which might, if not corrected for, be erroneously taken as mercury absorption.

In addition to the acid used, the length of the digestion period may affect the amount of mercury extracted. From the results given in Table 5 it is seen that an increase in the digestion time does not increase the amount of mercury found, but it improves the precision of replicate analyses of a sample.

TABLE 7

Comparison of Hg in area studied with other regions of Mediterranean.

| and the second section of the section of the second section of the second section of the second section of the section of the second section of the | August of the migration of the second | Organisms (µg kg <sup>-1</sup> FW) |                | Sediment        | Reference   |
|--|---------------------------------------|------------------------------------|----------------|-----------------|---|
| Water (ng l1)  |                                       | Family Mullidae Family Clupeida    |                | (µg kg-1 DW)    | Keterence   |
| urkish coast<br>his work)  | 15 (8-25)                             | 40<br>(20–80)                      | 50<br>(40–60)  | 140<br>(28–460) | _   |
| rael Coast   | 60<br>(10–180)                        | 310<br>(50–560)                    | 76<br>(30–120) | 130<br>(10-570) | Roth & Hornung, 1975  |
| Gibraltar  | 110<br>(60-110)                       |                                    | -              |                 | Robertson et al., 1972  |
| off Cyprus   | 120                                   | -                                  |                | -               | Robertson et al., 1972  |
| Fuscany coast  | (90–140)<br>(< 20–200)                | _                                  | _              | (40-1300)       | Renzoni et al., 1973  |
| S. W. Med.   | 24                                    |                                    | Ден            | · Care          | Fukai & Nyunh-Ngoc, 1976  |
| W. Ligurian Sea  | (16-30)<br>21<br>(17-30)              | _                                  |                |                 | Fukai & Hyunh-Ngoc, 1976  |
| Alboran Sea  | -                                     |                                    | _              | 260             | Robertson et al., 1972  |
| Tyrrhenian Sea   | 25                                    |                                    | ~~             | (50-1570)       | Fukai & Hyund-Ngoc, 1976 (water)<br>Selli et al., 1973 (sediment) |
| Off Pescara  | (20–30)                               | 100<br>(55–143)                    | -              | ***             | Caracciolo et al., 1972   |
| Carrilles Carls  | _                                     | (000000)                           | -              | 500             | Grimanis et al., 1976   |
| Saronikos Gulf Piraeus Harbor  |                                       | Table 1                            |                | 10 000          | Grimanis et al., 1976   |
| Off Monte Carlo  |                                       |                                    | -              | 300             | Robertson et al., 1972  |
| Off Marseilles   | AND                                   | . Interes                          | 205            |                 | Thibaud, 1971   |
| Off Sète   |                                       | need.                              | 235            |                 | Thibaud, 1971   |

The analytical results for the sediments taken from four locations are presented in Table 6. The mercury levels in all samples, with the exception of those from Mersin harbour, are low. The Mersin harbour sediments show elevated mercury levels, in comparison with nearby regions, possibly as a result of the extensive shipping activities.

## Conclusions

The average results obtained in this work are presented in Table 7 together with values reported from other regions of the Mediterranean. The low concentrations of mercury found in general in all samples from the eastern Mediterranean area studied support the view that as yet this region is relatively unpolluted. However, the higher concentration of mercury in the water and sediment samples from Mersin harbour, compared to nearby areas, demonstrate the effects of man's activities. Considering the Mediterranean as a whole, the mercury levels in this region are lower than other regions of the eastern Mediterranean; and in comparison with the western regions which have been most extensively studied, the difference is more pronounced. It is believed that the data presented provide some information as to the natural background levels of mercury in this particular region, and perhaps the Meliterrapean as a whole.

This study was performed as part of the joint FAO/UNEP pilot project on baseline studies and monitoring of metals, particularly mercury, in Mediterranean marine organisms, and as part of a project supported by the Turkish Electrical Authority. All assistance is gratefully acknowledged.

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